

# A Review Article of Nanoparticles; Synthetic Approaches and Wastewater Treatment Methods

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**Abstract:-** In Modern century the water demand is most facing challenge in the world, resulting from global progress and weather variation, require modern knowledge and methods to ensure the clean water safety and its supply for drinking. The adaptation of new and novel nanotechnology offers the opportunities to ensure the water purification process. The growth of cost and stable resources for supplying the drinking water at suitable level is the basic requirement of industries. Pollutant water treatment tools is unproductive for supplying sufficient clean water because the growing demand are linked with severe different health recommendations and impurities. The highly resourceful methods are giving the reasonable mixture to pollutant water assessment that does not depend on compacted systems. These have applications of the produced nano scales particles as effective adsorbents agent for eliminating the contaminated type material. The membrane separation tools incorporated with nanoparticles display a composite photo catalytic nature membrane containing vast potential to treat the different organic toxins in effluents. There were lot minerals, agro and clays and wastes have role in eliminating the metal waste product from the industrial effluents and natural water due to their ecological aspects. This conceptual review show current improvements in nanotechnology field special in the water treatment by using different nanomaterials, their fundamental properties, application, and mechanisms; as their benefits and limitations compared to present methods, few emerging tasks and some more quality base research require for their commercialization.

**Key Words:** Waste water treatment, TiO<sub>2</sub> material, Nanoparticles, Nano Catalyst, Nano Rode, Pollutants

## 1. INTRODUCTION

The water is the dynamic and energetic element in human life. Recently, the reported statement is that the world population suffers from the fresh/surface drinking water. So, the world has severe challenges regarding to the pure water because the accessible supplies of the water are reducing due to many factors such as population growth, extended droughts and competing burdens from several users [1-3]. So there is an urgent need to develop a novel technology to fulfill the affordable and clean water requirement for human being. The healthy body need clean and germ free water to maintain the life cycle. In some countries like Pakistan, Bangladesh and India there are water born disease; special drinking water. Recently, scientists are using several

methods for the water treatment to make it healthy for consumers. The commonly methods are physical and chemical such as chlorine and its related derivatives treatment, ultrasonic irradiation [4], reverse osmosis, ultraviolet light [5], boiling method, activated carbon based treatment etc. Today the nanotechnology is growing rapidly due to unique physical and chemical features of different nanomaterial. This type of technology increased a fantastic incentive due to their ability of conversion of the different metal into small sized. The Nano science called the study about the small things and the nanomaterial have large surface area to its volume as compared to other bulk material which helpful in catalysis [6-8] and medicine [9-13]. A nano range meter is approximately one billion and it denoted by size of the ten H-atoms are in straight line. Nanotechnology is manipulating matter at the very nanoscale level (down to 1/100,000 the width of a hair of human) in order to create progressive materials and efficient products with great potential to change the world [14]. It has ability to control and operate individual atoms and molecules. It has interdisciplinary nature, such as in the field of chemistry, physics, geothermal study, biology, and material science. The nanomaterial size is always less than 100nm (Figure 1). It contains high surface areas and adsorption characteristics due to having the diverse reactive surface sites and random surface regions [15]. The method that a crack produces at high scale, the bulk substances is not similar to the crack propagation as where particle size and crack are both comparable like Choi et al. [16] study different application of different scientific methods to fabricate nano shaped titanium oxide (TiO<sub>2</sub>) catalysts material. This kind of product can be useful in the photo catalytic procedure with novel features of great porosity (46%) and surface area (147 m<sup>2</sup>/g), pore-size distribution range is from 2- 8 nm, homogeneity without pinholes and cracks, minute crystallite size (9nm) and active crystal phase for improvement in water treatment. These kinds of photocatalysts of TiO<sub>2</sub> were proficient for the usage of dyes. Nanotechnology was invented by the Taniguchi (1974) as "Nano-technology generally consists of the consolidation, separation, processing and deformation of the materials by single atom or molecule" [17]. Generally, nanotechnology tools are assessing, modeling, imaging, and the manipulating the substance at small scale. The nanoparticles has vital role to the modification, production, and shape of the different structures which are playing role in manufacturing zone like in environmental applications [18-20].

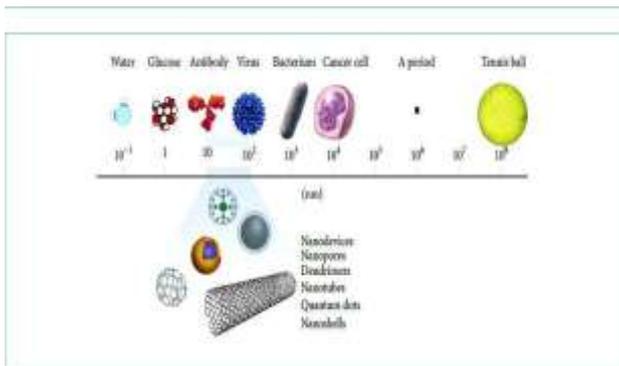


Figure 1; Different Size of Nanoparticle [67].

The impurity in water contains toxic metal like Ag(I), Ni(II), As(V), Cu(II), As(III), Hg(II) is creating an ecological problems [21]. To attain the ecological detoxification, several techniques like ion exchange, reverse osmosis, precipitation, flotation, electrochemical mechanism, oxidation processes and evaporation are commonly used. The nanostructured tools such as carbon nanotubes, C/Fe nanoparticles, magnetic nanoparticle, small structured iron zeolite, silver cyclodextrin composites, photo catalytic nanoparticles, nano filtration base membranes and active silica nanoparticles can be used to eliminate the toxic metals, some chemical discharges, sediments, charged based nanoparticles, some bacterial species and dangerous pathogens. [22]. Scientists categorized the nanoscale materials like purposeful materials for purification like metal based nanoparticles, dendrimers, zeolites and carbon base nanomaterial [23]. These can be effectively worked to the purification and treatment of water. Recently, the emerging field nanotechnology is best source to solve the water challenges (quality & quantity) around the world [24]. The nanomaterial e.g. carbon nanotubes (CNTs) and another called dendrimers are more effective treatment processes than other new methods because they have the remarkable adsorption characters [25-26]. There were a lot methods reported to solve different issues of water to ensure its safety and constancy as shown in table.

Table 1: Some Nanoparticles with their application

Nanoparticles	Nature of Removed Pollutant
Nano catalyst	Different Pesticides Azodyes, etc
Bioactive nanoparticle	Removal of Bacteria, fungi
Carbon nano tubes(CNT)	Organic Pollutant
Nano Structural catalytic	Organic pollutants

Different opportunities and tasks of using the nano sized materials in the purification and treatment of the surface, ground and industrial waste-water; the streams are problem as concern. The nanotechnology is helpful in curing diseases like renovating body tissue [27], killing toxic organisms [28].

## 2. SYNTHETIC APPROACHES FOR NANOPARTICLES

The nano materials are unstable in the nature, its production practices are essential for the practical uses. There were several systems are present to yield different type of nanomaterial. It can be formed with the help of larger structures (top down approach as shown in figure 2) in the presence of ultrafine grinder's technique or vaporization and lasers method. For complex and composite material, scientists usually choose to produce the nanostructures with the help of bottom-up method assembling it to the make complex arrangements with valuable properties. 'Layer by layer' based deposition technique is; the double layer membranes may be helpful for the analysis of the protein that produced through the layer of sodium based silicate and poly (allylamine hydrochloride) on metal i.e. calcinations of gold by heating system. The lipid double layer might be linked to silicate layers and it helpful identification of the proteins [29]. Rivero et al. [30] stated production of nanoparticles with diverse shape, size and dye (orange, violet, green) effective integration into the polyelectrolyte multi-layer films by using the layer-by-layer (LbL) method.

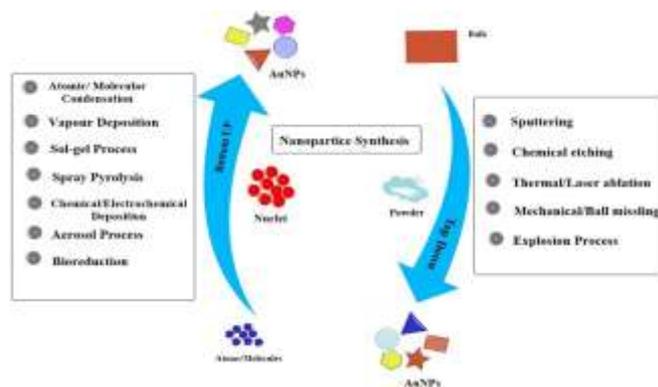


Figure 2; Synthesis approaches of Nanoparticles [68]

Zhao et al. [31-32] made film with multi-layer which contain the nanoparticles of silver and polycation poly (diallyldimethylammonium chloride) (PDDA) following analogous systems. The nanoparticle has fundamental role in the maintenance of temporal stability during the biocompatible fibers production. During that specific synthetic procedure crystallization is effective strategy. The nano particles & rods of gold with another shape were formed by using dead oat by incubation with aqueous solution of gold ions [33]. The gold or silver nanoparticles formed by using the fungi and bacteria [34]. The gas phase and sol-gel scheme are major synthesis methods belong to nanoparticles. Nanoparticles having diameters range from 1-10 nm with surface derivatization, crystal structure, and containing highly monodispersity and it treated with the gas-phase method and other effective sol-gel method. The fresh crystalline substance growth was relying on smaller size particles produced through the condensation or evaporation in inert-gas environment [35-36]. Several aerosol systems stated to develop a better fabrication of the nanoparticles

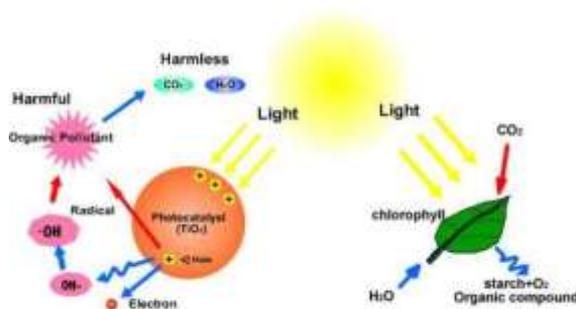
[37-38]. The self-assembly method, influence of the chemical and physical conditions like pH, solute concentrations and temperature can be induce the molecules to another fibrous nanostructures [39]. The 'Polymerosomes' is superior kind of the nanomaterial containing the immense potential in the perspective of waste water purification. The hydrophobic and hydrophilic groups present in polymerosomes produce a layer on copolymers that is useful in absorbent diversity of different molecules at varied pH values [40]. However, these copolymers are used for the treatment of toxic material from the different nature of water bodies.

### 3. WATER TREATMENT METHODS

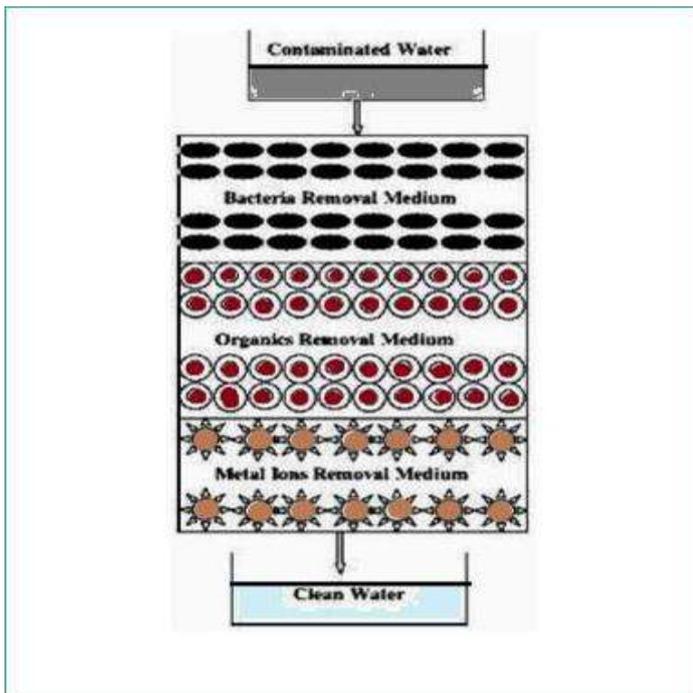
This unit reveals most capable information of technology where nanomaterial components serve as water treatment. The adsorption method is very helpful technique for the removal of toxic waste from water. Use of different type nanoparticles as an adsorbent in purification of the various toxic water in several forms such as nano composite, catalytic membrane, bioactive nanoparticles, catalytic, biomimetic membrane, polymeric thin film composite etc. The some organic compounds are absorbed with the help of CNT than actuated morphology based carbon [41]. The organic chemical containing hydroxyl group, amide and carboxylic groups form the H- bond with carbon nanotubes surface [42]. The commonly CNT contain greater adsorption level of metal ions [43-45]. The nano level metal oxides such as  $Al_2O_3$ , iron oxides,  $TiO_2$ , are low cost adsorbents for the radio nucleides and heavy metals [46-48]. Dendrimers is a polymeric nanomaterial which is able to eliminate the organics wastes & some heavy metals [49]. The smaller size membranes are a specific membranes modified with the nano fibers which used to eliminate micro size units with greater exclusion rate and having lower fouling tendency from aqueous medium phase [50]. Recent studies shows that nanotechnology has motivated in producing multifunction film through using nanomaterials like the inorganic and polymeric based membranes called the nanocomposite membranes. Addition of the metal based different oxide nanoparticles (silica, alumina and  $TiO_2$ ) to specific polymeric films has been presented to natural membrane fouling resistance, water absorptivity and surface hydrophilicity [51].

The nano- $TiO_2$  inorganic nature membranes or improved form of nano $TiO_2$  is applicable efficiently for the deprivation of pollutants like chlorinated based compounds [52]. The  $TiO_2$  applications on polyethylene and slurry of  $TiO_2$  mixture has verified by polymer based membrane for degradation in pharmaceuticals respectively [53-54]. In shortly the  $TiO_2$  is effective material to eliminate the pollutants as shown in figure 3. The  $Fe_2O_3$  and  $TiO_2$  nano composite is incorporated inside the ultrafiltration membranes effectively lower the polluting load and enhanced permeate flux [55]. Alumina-zirconia-titania is clay base membrane layered with compound  $Fe_2O_3$  particles was experimentally lower carbon well than non-coated material membrane increasing the degradation of the natural containing organic matter [56-57]. So, the ceramic nanocomposite of CNTs &  $TiO_2$  enhanced the membrane permeability and its photo catalytic activity [58-60]. Antimicrobial nanomaterials like small sized silver are fixed to prevent the attachment of bacterial and its development on the surface of biofilm [61-62]. It deactivates the viruses and commonly they can lower the bio-fouling nature [63]. Growths of thin layer base nanomaterial mostly focus on uniting the materials into the thin film nanocomposite via doping on surface modification. The membrane permeability effect of nanomaterial, its selectivity focus on the variability, quantity & dimension of nanoparticles [64-65]. The application of nano fibrous based composites having membranes for the organic waste treatment is partial and other stand-alone system (Figure 4) planned for eliminating whole kinds of pollutants containing bacteria, heavy metals, viruses and impure ions. Nano catalysts are also very important role in removing the impurities from the water due to their remarkable aspects. Because the presence of greater surface volume area and their morphology, the zero-valent metal, effective semiconductor base materials and other its bimetallic nanoparticles are considered in water purification as raise catalytic movement on the surface. The degradation and reactivity increase of different environmental toxins such as organo chlorine, herbicides etc.

Silver nanoparticles might be produced through using bacteria named; *Bacillus cereus*. This strain showing diverse absorptions of the silver salt deliberates with analytical tools i.e. Energy Dispersive spectroscopy (EDS), High Resolution Transmission Electron Micrography (HRTEM) and X-ray diffraction (XRD). Prakash et al. [66] have described Cellulose acetate and MgO nano based particles. The fibers of cellulose acetate surrounded with nanoparticles of Ag against the spores and bacteria.



**Figure 3;** The Removal of Pollutants by using  $TiO_2$  material [69]



**Figure 4;** Schematic scheme of Nano fibrous media and membrane filters to remove the pollutants from the wastewater [70]

#### 4. CONCLUSION

The field nanotechnology is deliberated to modern period, material linked to subject essentially unidentified to the folk's due to its modern technology. Very soon the nano sized type materials approaches have play significant role at large scale to purify the water from pollutants. Therefore nanoparticles were dominant process to reduce the toxin metal from water with great efficiency. Metal nanoparticles are appropriate to remove several toxin metals such as Arsenic (As). The metal nanoparticles are cheap and easy to run and control, with cheerful future for its further application. The metal nanoparticles properties report the challenges in many countryside areas where are absence of different resources and suitable technology for the water purification. This is generally for minor scale water purification systems. It aiding between 500-1000 people and is an emerging ideal developing the technology to make availability for unpolluted and germ free water to rural zones.

#### REFERENCES

1. US Bureau of Reclamation and Sandia National Laboratories. Desalination and water purification technology roadmap a report of the executive committee Water Purification. 2003.
2. US Environmental Protection Agency. Microbial and disinfection by-product rules. Federal Register. 1998; 63: 69389-69476.

3. US Environmental Protection Agency. Alternative disinfectants and oxidants guidance manual. EPA Office of Water Report. 1999; 815: 99-114.
4. Gupta S, Behari J and Kesari K. Low frequency ultrasonic treatment of sludge. Asian Journal of Water Environment and Pollution. 2006; 3: 101-105.
5. Droste RL. Theory and practice of water and wastewater treatment. New York: Wiley (Book). 1997.
6. Shin, T.-Y.; Yoo, S.-H.; Park, S., Gold nanotubes with a nanoporous wall: their ultrathin platinum coating and superior electrocatalytic activity toward methanol oxidation. *Chemistry of Materials* **2008**, *20* (17), 5682-5686.
7. Xian, Y.; Gao, F.; Cai, B., Synthesis of platinum nanoparticle chains based  $\alpha$ -chymotrypsin fibrils. *Materials Letters* **2013**, *111*, 39-42.
8. Ge, J.; Huynh, T.; Hu, Y.; Yin, Y., Hierarchical magnetite/silica nanoassemblies as magnetically recoverable catalyst-supports. *Nano letters* **2008**, *8* (3), 931-934.
9. (Hamouda, I. M., Current perspectives of nanoparticles in medical and dental biomaterials. *Journal of biomedical research* **2012**, *26* (3), 143-151;
10. Albrecht, M.; Janke, V.; Sievers, S.; Siegner, U.; Schöler, D.; Heyen, U., Scanning force microscopy study of biogenic nanoparticles for medical applications. *Journal of magnetism and magnetic materials* **2005**, *290*, 269271.
11. Rai, M.; Yadav, A.; Gade, A., Silver nanoparticles as a new generation of antimicrobials. *Biotechnology advances* **2009**, *27* (1), 76-83.
12. Piao, M. J.; Kang, K. A.; Lee, I. K.; Kim, H. S.; Kim, S.; Choi, J. Y.; Choi, J.; Hyun, J. W., Silver nanoparticles induce oxidative cell damage in human liver cells through inhibition of reduced glutathione and induction of mitochondria-involved apoptosis. *Toxicology letters* **2011**, *201* (1), 92-100.
13. Chen, X.; Schluesener, H., Nanosilver: a nanoparticle in medical application. *Toxicology letters* **2008**, *176* (1), 1-12.
14. Monika, T., Dhruv, T. Polymer nanocomposite and their application in electronics industry. *Int. J. Electron. Elec. Eng.*, **7**(6), 603-608(2014).

15. Ichinose N, Ozaki Y and Kashu S. Superfine particle technology. Springer, London, (Book). 1992.
16. Choi H, Stathatos E and Dionysiou, DD. Sol-gel preparation of mesoporous photocatalytic TiO<sub>2</sub> films and TiO<sub>2</sub>/Al<sub>2</sub>O<sub>3</sub> composite membranes for environmental applications. *Applied Catalysis B: Environmental*. 2006; 63: 60-67.
17. Taniguchi N. On the Basis Concept of "Nanotechnology", Proceeding of the International Conference on Precision Engineering, Tokyo part II, Japan Society of Precision Engineering. 1974.
18. Lunge SS, Singh S and Sinha A. International Conference on Chemical, Civil and Environmental Engineering (CCEE'2014), Magnetic Nanoparticle: Synthesis and Environmental Applications, Singapore. 2014; 18-20.
19. Bhattacharya S, Saha I, Mukhopadhyay A, et al. Role of nanotechnology in water treatment and purification: Potential applications and implications. *International Journal of Chemical Science and Technology*. 2013; 3: 59-64.
20. Seitz F, Pollmann K, Mackenzie K, et al. Photo oxidation in Combination with Nanotechnologies Principles, Developments and R&D Approaches of an Advanced Technology For Water and Air Treatment- Uviblox®. *Journal of Advanced Oxidation Technology*. 2011; 14: 260-266.
21. Friberg L, Nordberg GF, and Vouk V.B. eds. *Handbook of Toxicology of Metals*; Elsevier: North-Holland. 1979; 355.
22. Nowack B. Pollution prevention n treatment using Nanotechnology. In *Environmental Aspects*. 2008.
23. Tiwari DK, Behari J and Sen P. Application of Nanoparticles in Waste Water Treatment. *World Applied Sciences Journal*. 2008; 3: 417-433.
24. Bottero J-Y, Rose J and Wiesner MR. Nanotechnologies: tools for sustainability in a new wave of water treatment processes. *Integrated Environmental Assessment and Management*. 2006; 2: 391-395.
25. Obare SO and Meyer GJ. Nanostructured materials for environmental remediation of organic contaminants in water. *Journal of Environmental Science and Health Part A*. 2004; 39: 2549-2582.
26. Savage N and Diallo MS. Nanomaterials and water purification: opportunities and challenge. *Journal of Nanoparticle Research*. 2005;7: 331-342.
27. Bindhu MR and Umadevi M. Synthesis of monodispersed silver nanoparticles using Hibiscus cannabinus leaf extract and its antimicrobial activity. *Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy*. 2013; 101: 184-190.
28. Walmsley GG, McArdle A, Tevlin R et al. *Nanotechnology in bone tissue engineering Nanomedicine*. 2015; 11: 1253-1263.
29. Phillips KS, Han JH, Martinez M, et al. Nanoscale glassification of gold substrates for surface plasmon resonance analysis of protein toxins with supported lipid membranes. *Analytical Chemistry*. 2006; 78: 596-603.
30. Rivero PJ, Goicoechea J, Urrutia A et al. Multicolor Layer-by-Layer films using weak polyelectrolyte assisted synthesis of silver nanoparticles *Nanoscale Research Letters*. 2013; 8: 438.
31. Zhao S, Zhang K, An J et al. Sun Synthesis and layer-by-layer self-assembly of silver nanoparticles capped by mercaptosulfonic acid *Materials Letters*. 2006; 60: 1215-1218.
32. Boanini EP, Torricelli M, Gazzano R et al. Nanocomposites of hydroxyapatite with aspartic acid and glutamic acid and their interaction with osteoblast-like cells. *Biomaterials*. 2006; 27: 4428-4433.
33. Henkel A, Schubert O, Plech A, et al. Growth Kinetic of a Rod-Shaped Metal Nanocrystal. *J Phys. Chem C*. 2009; 113: 10390-10394.
34. Vahabi K, Mansoori GA and Karimi S. Biosynthesis of silver nanoparticles by fungus *Trichoderma reesei* (a route for largescale production of AgNPs). *Insciences Journal*. 2011; 1: 65-79.
35. Siegel RW. *Nanomaterials: synthesis, properties and applications Annual Review of Materials Science*. 1991; 21: 559.
36. Siegel RW. *Physics of new materials*. Fujita, F.E. (Ed.). Springer Series in Materials Science, 27: Berlin: Springer. 1994.
37. Uyeda R. *Studies of ultrafine particles in Japan: Crystallography, methods of preparation and technological applications*. Materials Science. 1991; 35: 1-96.
38. Friedlander SK. Synthesis of nanoparticles and their agglomerates: Aerosol reactors. In *R and D status and trends*, ed. Siegel et al. 1998.

39. Graveland BJF and Kruif CG. Unique milk protein based nanotubes: Food and nanotechnology meet. *Trends Food Science Technology*. 2006; 17: 196-203.
40. Lorenceau E, Utada AS, Link DR et al. G. Generation of polymerosomes from double-emulsions *Langmuir*. 2005; 21: 9183-9186.
41. Prachi P, Gautam D, Madathi, Nair ANB. Nanotechnology in Waste Water Treatment: A Review, *International Journal of Chemical Technology Research*. 2013; 5: 2303-2308.
42. Pan B and Xing BS. Adsorption mechanisms of organic chemicals on carbon nanotubes, *Environmental Science and Technology*. 2008; 42: 9005-9013.
43. Yang K, Wu WH, Jing QF et al. Aqueous adsorption of aniline, phenol, and their substitutes by multi-walled carbon nanotubes, *Environmental Science and Technology*. 2008; 42: 7931-7936.
44. Rao GP, Lu C and Su F. Sorption of divalent metal ions from aqueous solution by carbon nanotubes: A review *Separation and Purification Technology*. 2007; 58: 224-231.
45. Li YH, Ding J, Luan ZK et al. Competitive adsorption of Pb<sup>2+</sup>, Cu<sup>2+</sup> and Cd<sup>2+</sup> ions from aqueous solutions by multiwalled carbon nanotubes *Carbon*. 2003; 2787-2792.
46. Lu CS, Chiu H and Liu CT. Removal of zinc II. from aqueous solution by purified carbon nanotubes: kinetics and equilibrium studies *Industrial and Engineering Chemistry Research*. 2006; 45: 2850-2855.
47. Deliyanni EA, Bakoyannakis DN, Zouboulis AI and Matis KA. Sorption of AsV. ions by akaganeite-type nanocrystals. *Chemosphere*. 2003; 50: 155-163.
48. Mayo JT, Yavuz C, Yean S, et al. The effect of nanocrystalline magnetite size on arsenic removal *Science and Technology of Advanced Materials*. 2007; 1-2.
49. Diallo MS, Christie S, Swaminathan P et al. Dendrimer enhanced ultrafiltration.1. Recovery of CuII. from aqueous solutions using PAMAM dendrimers with ethylenediamine core and terminal NH<sub>2</sub> groups. *Environmental Science and Technology*. 2005; 39: 1366-1377.
50. Ramakrishna S, Fujihara K, Teo WE, et al. Electrospun nano fibers: solving global issues. *Materials Today*. 2006; 9: 40-50.
51. Maximous N, Nakhla G, Wong K and Wan W. Optimization of Al<sub>2</sub>O<sub>3</sub>/PES membranes for wastewater filtration *Separation and Purification Technology*. 2010; 73: 294-301.
52. Wu L and Ritchie SMC. Enhanced dechlorination of trichloroethylene by membrane-supported Pd-coated iron nanoparticles *Environmental Progress*. 2008; 27: 218-224.
53. Lin HF, Ravikrishna R and Valsaraj KT. Reusable adsorbents for dilute solution separation. 6. Batch and continuous reactors for the adsorption and degradation of 1,2-dichlorobenzene from dilute wastewater streams using titania as a photocatalyst. *Separation and Purification Technology*. 2002; 28: 87-102.
54. Molinari R, Palmisano L, Drioli E and Schiavello M. Studies on various reactor configurations for coupling photocatalysis and membrane processes in water purification. *Journal of Membrane Science*. 2002; 206: 399-415.
55. Sun D, Meng TT, Loong TH and Hwa TJ. Removal of natural organic matter from water using a nano-structured photocatalyst coupled with filtration membrane. *Water Science and Technology*. 2004; 49:103-110.
56. Karnik BS, Davies SH, Baumann MJ and Masten SJ. Fabrication of catalytic membranes for the treatment of drinking water using combined ozonation and ultrafiltration. *Environmental Science and Technology*. 2005; 39: 7656-7661.
57. Karnik BS, Davies SH, Chen KC et al. 2005. Effects of ozonation on the permeate flux of nanocrystalline ceramic membranes. *Water Research*. 2005; 39: 728-734.
58. Verweij H, Schillo MC and Li J. Fast mass transport through carbon nanotube membranes. *Small*. 2007; 3: 1996-2004.
59. Yang GCC and Tsai C.-M. Preparation of carbon fibers/carbon/alumina tubular composite membranes and their applications in treating Cu-CMP wastewater by a novel electrochemical process. *Journal of Membrane Science*. 2008; 321: 232-239.
60. Yao Y, Li G, Gray KA, and Lueptow RM. Single-walled carbon nanotube-facilitated dispersion of particulate TiO<sub>2</sub> on ZrO<sub>2</sub> ceramic membrane filters. *Langmuir*. 2008; 24: 7072-7075.
61. Mauter MS, Wang Y, Okemgbo KC, et al. Antifouling ultrafiltration membranes via post-fabrication

- grafting of biocidal nanomaterials. *Applied Materials and Interfaces*. 2011; 3: 2861-2868.
62. Zodrow K, Brunet L, Mahendra S, et al. Polysulfone ultrafiltration membranes impregnated with silver nanoparticles show improved biofouling resistance and virus removal. *Water Research*. 2009; 43: 715-723.
63. De Gusseme B, Hennebel T, Christiaens E et al. Virus disinfection in water by biogenic silver immobilized in polyvinylidene fluoride membranes. *Water Research*. 2011; 45: 1856-1864.
64. Jeong BH, Hoek EMV, Yan YS et al. Interfacial polymerization of thin film nanocomposites: a new concept for reverse osmosis membranes. *Journal of Membrane Science*. 2007; 294: 1-7.
65. Lind ML, Ghosh AK, Jawor A et al. Influence of zeolite crystal size on zeolite polyamide thin film nanocomposite membranes. *Langmuir*. 2009; 25: 10139-10145.
66. Prakash S, Sharma N, Ahmad A and Ghosh P. Synthesis of Agnps By Bacilus Cereus Bacteria and Their Antimicrobial Potential. *Journal of Biomaterials and Nanobiotechnology*. 2011; 2: 15-16.
67. Amin MT, Alazba AA, and Manzoor U. A Review of Removal of Pollutants from Water/Wastewater Using Different Types of Nanomaterials. *Advances in Materials Science and Engineering*. 2014; 1-24.
68. Dr. Suparna Mukherji, Ms. Sharda Bharti, Dr. Soumyo, Ms, Gauri; Synthesis and characterization of size- and shape-controlled silver nanoparticles, 2018-08-25 | DOI: <https://doi.org/10.1515/psr-2017-0082>
69. Hala Elia, Using Nano- and Micro-Titanium Dioxide (TiO<sub>2</sub>) in Concrete to Reduce Air Pollution, *J Nanomed Nanotechnol* 2018; 9:505, Vol 9(3) DOI: 10.4172/2157-7439.1000505
70. Mamadou SD and Savage N. Nanoparticles and water quality. *Journal of Nano Research*. 2005; 7: 325-330. <http://dx.doi.org/10.1016/j.jare.2015.02.007>, 2090-1232<sup>a</sup> 2015 Production and hosting by Elsevier B.V. on behalf of Cairo University.