**ASSESSMENT OF QUALITY OF DRINKING WATER IN DIFFERENT ZONES OF DELHI**

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Abstract - The report outlines the water supply of drinking water and comparison of parameters of drinking water in various zones of Delhi as per defined in the IS code 10500: 2012, World Health Organization (WHO), Delhi Jal Board (DJB). Seven tests are performed taking up the physical, chemical, biological parameters into consideration. The parameters taken up provide a brief idea whether the quality of drinking water is potable or not and safe to drink or not. In this project we have conducted several tests on water samples collected from different zones of Delhi and observed that which zone of Delhi is under below the level of safety for drinking water and which zone of Delhi is safest for drinking water. We have also observed the overall average readings in form of graphs and result analysis.

Key Words: Water quality Index, Various meters, Turbidity, pH, DO, TDS etc.

1. INTRODUCTION

Drinking water is also known as potable water or better drinking water or intake able water which is safe to drink or to use for food preparation, without risk of health problems. In Delhi 2017, 90% of people had access to water suitable for drinking. Many consumers had access to tap water while another had access to wells or public taps or other water supplies. There is still use an unsafe drinking water source which may be contaminated by feces. Water with Contaminants can lead to various issues like:- Diarrhea, cholera, Typhoid, Psychological effects in human system, Blood Pressure, Less killing of microorganism. Diseases caused by Water Viral Infections like:- Polio virus, Infectious jaundice. It particularly leads in young children and pregnant women.

Our aim is to do this research paper that:

- To observe and test the parameters for drinking water as per IS 10500: 2012 in different zones of Delhi.
- Depending upon the test quality of water can be intake by the peoples or consumers is suitable for drinking or not.
- To determine that how does potable water relate to our health in different zones.
- How to overcome with diseases and what are their remedies.
- Whether there is a need of RO or Aquaguards or other water Purifiers.
- Zone 1 is North Zone, Zone 2 is West Zone, Zone 3 is East Zone, Zone 4 is South Zone.

2. MATERIALS AND METHOD

The physical, chemical and biological parameters are as follows:-

i) Turbidity

It is the muddiness or lack of clarity of a water sample caused by large numbers of individual particles that are generally invisible to the naked eye, similar to smoke in air. The measurement of turbidity is an explanation test of water quality. A meter called Turbidimeters which is used as a water cleansing plant instrument to find out turbidity (in NTU) of untreated water and clear water after filtration.

ii) pH value

It is a measure of the hydrogen ion concentration of a solution. Solutions having high concentration of hydrogen ions have low pH and solution having low concentrations of H⁺ ions have a high pH value.

iii) Hardness

Water hardness is the conventional measure of the capacity of water to react with soap, hard water require very much more soap to produce lather. Hardness is most commonly expressed as milligrams (mg) of calcium carbonate equivalent (CaCO₃) per litre. Water containing calcium carbonate (CaCO₃) at concentrations below 60 mg/l is generally considered as soft; 60–120 mg/l, moderately hard; 120–180 mg/l, hard; and more than 180 mg/l, very hard. While hardness is cause by cations, it can also be discussed in...
terms of carbonates (Temporary Hardness) and non-carbonates (Permanent Hardness).

iv) Total Solids & TDS (Total Dissolved Solids)
It is a measure of the suspended and dissolved solids in water. Suspended solids are those that can be retained on a water filter and are capable of settling out of the water column onto the stream bottom when stream velocities are low. Dissolved solids refer to any inorganic salts, small amount of organic matter and dissolved gases. It is computed by TDS meter.

v) DO (Dissolved Oxygen)
It is the quantity of gaseous oxygen (O₂) dissolved in the water. Oxygen enters the water by direct absorption from the atmosphere, by rapid movement, or as a waste product of plant photosynthesis. Water temperature and the volume of moving water can affect dissolved oxygen levels.

3. TESTING PROCEDURES

i) Residual Chlorine test
Chlorides are found to occur in all the natural waters, and their quantity may vary widely. Chlorides in reasonable concentrations are not harmful for humans i.e. 0.2-0.3 mg/l.

ii) Hardness
The hardness usually caused by the presence of calcium and magnesium salts present in water.

iii) pH
It is the log of a reciprocal of hydrogen ions present in that water i.e. pH = log₁₀ X 1/H⁺. It is measured by pH meter.

iv) Dissolved Oxygen
The amount of oxygen found dissolved in a given water at a given temperature and pressure. It is measured by DO meter.

v) Turbidimeter
Turbidity is caused in drinking water by divided suspended particles of clay, silt, sand or by organic materials. The permissible turbidity for drinking water is 5 to 10 units or below 5 units. It is measured by turbidity meter.

vi) TDS (Total Dissolved Solids)
It consists of mainly inorganic salts, small amount of organic matter and dissolved gases. It is computed by TDS meter.

vii) Total solids
These can be found by filtering the water sample and weighing the residue left on the filter paper. The total solids (T.S) in a given water up to 500 mg/l.

It is calculated by concentration of Total solids = (M₂ - M₁ / 50 x 1000).

Where,
Mass of empty dish = M₁ gm
Mass of filter paper + dry residue = M₂ gm
Mass of filter paper + dry residue = M₃ gm

After getting readings we will compare the results with drinking water as per IS 10500: 2012, Delhi Jal Board, WHO.

4. OVERALL RESULTS & COMPARISON

<table>
<thead>
<tr>
<th>Test Name</th>
<th>Our Observation</th>
<th>As per IS 10500:2012</th>
<th>IS WHO</th>
<th>As per DJB</th>
</tr>
</thead>
<tbody>
<tr>
<td>pH</td>
<td>7.79</td>
<td>6.5-8.5</td>
<td>6.5-8</td>
<td>6.5-8</td>
</tr>
<tr>
<td>Turbidity</td>
<td>0.06</td>
<td>MAXIMUM 1 NTU</td>
<td>1-5</td>
<td>0.5-1</td>
</tr>
<tr>
<td>DO (Dissolved Oxygen)</td>
<td>7.02</td>
<td>6-8ppm</td>
<td>5-7ppm</td>
<td>6-8ppm</td>
</tr>
<tr>
<td>Residual Chlorine</td>
<td>0.04</td>
<td>0.2-1 mg/l</td>
<td>0.2-1 mg/l</td>
<td>0.20-0.50</td>
</tr>
<tr>
<td>Hardness</td>
<td>187</td>
<td>200-600 mg/l</td>
<td>75-115 mg/l</td>
<td>300-600</td>
</tr>
<tr>
<td>TDS</td>
<td>179.8</td>
<td>500-2000 mg/l</td>
<td>500-1500 mg/l</td>
<td>500-1000</td>
</tr>
<tr>
<td>TS (Total Solids)</td>
<td>242.5</td>
<td>500-2000 mg/l</td>
<td>500-1500 mg/l</td>
<td>500-1000</td>
</tr>
</tbody>
</table>

**Table 2 Zone 2**

<table>
<thead>
<tr>
<th>Test Name</th>
<th>Our Observation</th>
<th>As per IS 10500:2012</th>
<th>IS WHO</th>
<th>As per DJB</th>
</tr>
</thead>
<tbody>
<tr>
<td>pH</td>
<td>7.26</td>
<td>6.5-8.5</td>
<td>6.5-8</td>
<td>6.5-8</td>
</tr>
</tbody>
</table>
### Turbidity
- **Maximum**: 1 NTU
- **Range**: 0.5 - 1

### DO (Dissolved Oxygen)
- **Range**: 5 - 8 ppm

### Residual Chlorine
- **Range**: 0.2 - 1 mg/l

### Hardness
- **Range**: 200 - 600 mg/l (safe drinking hardness is between 75 - 115 mg/l)

### TDS
- **Range**: 157.2 - 500 mg/l
- **Range**: 500 - 1000 mg/l

### TS (Total Solids)
- **Range**: 208.4 - 500 mg/l
- **Range**: 500 - 1000 mg/l

### Table 3 Zone 3

<table>
<thead>
<tr>
<th>Test Name</th>
<th>Our Observation</th>
<th>As per IS 10500:2012</th>
<th>As per WHO</th>
<th>As per DJB</th>
</tr>
</thead>
<tbody>
<tr>
<td>pH</td>
<td>7.20</td>
<td>6.5 - 8.5</td>
<td>6.5 - 8.5</td>
<td>6.5 - 8.5</td>
</tr>
<tr>
<td>Turbidity</td>
<td>1.7</td>
<td>MAXIMUM 1 NTU</td>
<td>1 - 5</td>
<td>0.5 - 1</td>
</tr>
<tr>
<td>DO (Dissolved Oxygen)</td>
<td>7.52</td>
<td>6 - 8 ppm</td>
<td>5 - 7 ppm</td>
<td>6 - 8 ppm</td>
</tr>
<tr>
<td>Residual Chlorine</td>
<td>0.02</td>
<td>0.2 - 1 mg/l</td>
<td>0.2 - 1 mg/l</td>
<td>0.20 - 0.50</td>
</tr>
<tr>
<td>Hardness</td>
<td>156</td>
<td>200 - 600 mg/l</td>
<td>75 - 110 mg/l</td>
<td>300 - 600 mg/l</td>
</tr>
<tr>
<td>TDS</td>
<td>251</td>
<td>500 - 2000 mg/l</td>
<td>500 - 1500</td>
<td>500 - 1000</td>
</tr>
<tr>
<td>TS (Total Solids)</td>
<td>341</td>
<td>500 - 2000 mg/l</td>
<td>500 - 1500</td>
<td>500 - 1000</td>
</tr>
</tbody>
</table>

### Table 4 Zone 4

<table>
<thead>
<tr>
<th>Test Name</th>
<th>Our Observation</th>
<th>As per IS 10500:2012</th>
<th>As per WHO</th>
<th>As per DJB</th>
</tr>
</thead>
<tbody>
<tr>
<td>pH</td>
<td>7.62</td>
<td>6.5 - 8.5</td>
<td>6.5 - 8.5</td>
<td>6.5 - 8.5</td>
</tr>
<tr>
<td>Turbidity</td>
<td>0.0</td>
<td>MAXIMUM 1 NTU</td>
<td>1 - 5</td>
<td>0.5 - 1</td>
</tr>
</tbody>
</table>

### Chart 1: pH of Different zones

**INTERPRATION OBTAINED**

1) **Lesser pH** can cause tuberculation and corrosion of pipes and treatment tanks.

2) **Higher pH** (alkaline) produce incrustation, sediment deposit, difficulty in chlorination, psychological effects on human systems.

3) **All Zones** are safe but few localities like Seelampur, Laxmi Nagar, Shahdara has less pH cause acidic in nature which harmful to humans and may cause tuberculation like diseases.

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4) Locality which may observe with high pH which is basic in nature may cause psychological effects on human systems.

2) Higher DO can cause corrosion in the supplying such Waters.

3) All Zones are safe but few localities like Seelampur, Laxmi Nagar, Shahdara has less pH cause acidic in nature which harmful to humans and may cause tuberculation like diseases.

INTERPRATION OBTAINED

1) Lesser DO can cause and reduce i.e. the organic pollution takes place with more generations of microorganisms formation and causes diseases.

2) Higher Residual chlorine can cause salty taste to water and causes high blood pressure and Heart attack.

3) All Zones are safe but few localities like Patel Nagar has more hardness or no residual chlorine which is unsafe for drinking because there is no purification and filtration of water takes place.
INTERPRATION OBTAINED

1) High contents may cause psychological affects and may also cause the growth of microorganisms which increases the infections and other water borne diseases.

2) Water with less content may cause diabetes and blood pressure fluctuations.

3) All zones is not very safe especially Zone 2 and Zone 4 because of less contents of TDS involved in it. It generally may cause diseases when migration takes place water parameters and the conditions may vary place to place.

2) Water with hardness upto 75ppm are considered as Soft and above 200ppm are considered as hard.

3) Zones2 is quite safe but few localities are not safe like Patel Nagar, Moti Nagar has more hardness.

5. CONCLUSIONS

Based on the Experimental result following points are summarized. To aware consumers that what parameters or kind of water they are drinking is it fit or not. On an overall average if we take of zones, East Zone is very harmful zone of drinking water, Secondly North Zone, West Zone and South Zone water is safe for drinking. Changes of quantities/parameters to be taken up in drinking water in all zones like North Zone Residual Chlorine is indicated to be less its value should be increase. West Zone – Hardness is more i.e. hard water supply takes place in this region can be balance. East Zone –Overall all parameters should be considered to check except pH because this zone is danger zone for drinking water. South Zone – Every parameter is similar as per codes and health organizations. But it is advised to drink water of moderate TDS because if migration takes place it causes diseases. From the above discussion it is identified that the use of toxic water can harm us. Excess of any quantity in water is bad/harmful as well as lack of any quantity in water is also bad/harmful. So it advised to make water potable in neutral parameter.

6. REFERENCES


