

A Basic study on Calcium Carbide Aerated Geopolymer with Pozzolanic Powder from fly ash and Rice Husk Ash

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Abstract - In this paper, a basic study is done on use of calcium carbide as aerator in Fly ash and Rice husk Ash Geopolymer. Geopolymer is made combining alkali activator solution of 10 moles and 12 moles of Sodium hydroxide (NaOH) and sodium silicate in 1:2.5 ratio. And Pozzolanic powder from mixing of 90% fly ash and 10 % Rice husk ash. Calcium carbide is added as percentage by weight of pozzolanic powder. Calcium reacts with water in alkali activator solution to form acetylene gas and residue of calcium hydroxide. The acetylene gases produced form bubbles, which act as aerator in geopolymer and reduces the density of geopolymer to 1300 kg/m³. The percentage water absorption is less than 14% for all samples.

Key Words: (Size 10 & Bold) Aerated concrete¹, Calcium carbide², geopolymer³, Sodium hydroxide⁴, sodium silicate⁵, Fly ash⁶, Rice Husk Ash⁷.

1. INTRODUCTION

1.1 Geopolymer

Geopolymers were invented by Davidovits [1] in 1970s. He used geopolymer as a term to describe inorganic materials with polymeric Si-O-Al bonds obtained from alumina-silicate oxide with alkali silicates. The network is made up with SiO₄ and AlO₄ tetrahedra linked alternately by sharing all oxygen atoms. The Al³⁺ in IV-fold coordination becomes a network forming but requires extra charge to compensate, which forces the presence of cations in the framework to balance the structure. According to Davidovits, the empirical formula of geopolymers or poly-sialates is as follows:



Geopolymers in this research is made from alkali activator viz. sodium hydroxide with sodium silicate solution and fly ash and Rice husk ash.

1.2 Aerated Concrete

N. Narayanan et. Al [2] have studied structure and properties of aerated concrete. In their study they suggest that Aerated concrete is relatively homogeneous when compared to normal concrete, as it does not contain coarse aggregate phase, yet shows vast variation in its properties. The properties of aerated concrete depend on its

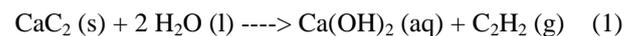
microstructure (void-paste system) and composition, which are influenced by the type of binder used, methods of pore-formation and curing. Although aerated concrete was initially envisaged as a good insulation material, there has been renewed interest in its structural characteristics in view of its lighter weight, savings in material and potential for large scale utilization of wastes like pulverized fuel ash (fly ash).

1.3 Aerated geopolymer

Zhitao Chen, et. Al [3] studied aerated geopolymer using aluminium powder. They suggested the introduction of gas in aerated concrete by the use of finely divided aluminum powder. The aluminum reacts with the soluble alkalis in the cement slurry to generate small bubbles of hydrogen. Their results showed that the utilization of IBA to synthesize aerated geopolymer with low density is feasible.

Normally aluminium powder is used as aerator, but in our study we have used calcium carbide as aerator. Calcium carbide residue which is nothing but calcium oxide is being widely studied as admixture in concrete. [4][5].

Calcium carbide normally reacts with water to produce heat and acetylene gas. This acetylene gas creates bubbles to create a aerated geopolymer. The reaction of calcium carbide with water is as follows:



Aluminium powder produces hydrogen gas while calcium carbide produces acetylene gas.

1.4 Rice Husk ash

B. Prabhu, et. Al [6], say that Rice husk ash (RHA) is a by-product from the burning of rice husk at a temperature lower than 600°C. This means that it is in a form that is soft and easy to grind. Rice husk ash is rich in silica about 90%.

Gemma Rodriguez de Sensale [7] concluded from her studies that RHAs in concrete reduces the mass loss of specimens exposed to hydrochloric acid solution and decreases the expansion due to sulfate attack and the alkali-silica reaction.

1.5 Fly ash

Fly Ash, an industrial waste obtained from Thermal Power Plants, its current annual production is 184.14 million tonnes in the year 2015 [8]. It is widely used in Portland pozzolana cement, high volume fly ash concrete, ready mix

concrete plants, hollow and solid concrete blocks manufacturing units, cement brick manufacturing plants, etc.

2. METHODOLOGY

The alkali activator solution is made in two sets as 10 mole and 12 mole solutions, keeping the ratio of sodium hydroxide to sodium silicate as 1:2.5. The solution is to be prepared one day prior to casting of cubes.

10% of Fly ash is partially replaced by Rice husk ash. Both fly ash and RHA are mixed thoroughly in dry powder form to get a pozzolanic powder.

The geopolymer paste was prepared by mixing alkali activator solution to the pozzolanic powder in 1:3 ratio.

Calcium carbide (CaC_2) was added in various percentages by weight of pozzolanic powder viz. 1%, 2%, 3%, 4% and 5% to the geopolymer paste.

In one set of 10 mole specimens 5% super plasticizer was added for all percentages of calcium carbide. Hence two types of 10 mole paste was created i.e. one without super plasticizer named as 10M and one with super plasticizer named as 10MS. Another set of 12 mole specimen without super plasticizer is named as 12M.

The geopolymer pastes were cast into cubes of 50 square centimeter surface area and cured in ambient temperature for 3 days.

The cubes are then tested for density, water absorption and compressive strength.



Fig -1: Pozzolanic Powder



Fig -2: Geopolymer Paste



Fig -3: Casting of cubes



Fig -4: Compression test on cubes



Fig -5: Broken cubes showing inner pores



Fig -6: Broken cubes fragments showing inner pores

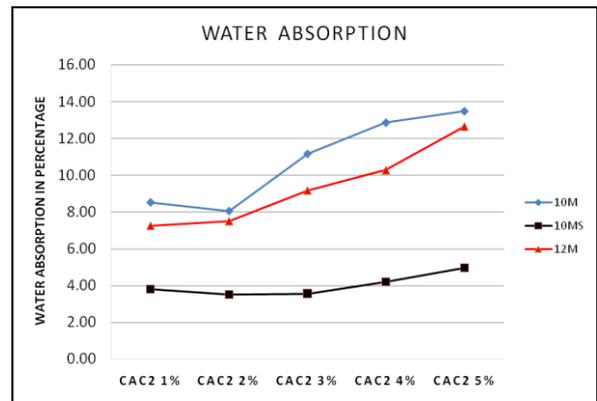


Chart -2: Water Absorption comparison

Table -1: Compressive strength comparison

MIX	Compressive strength in N/mm ²		
	10M	10MS	12M
CaC2 1%	1.9	2.1	1.2
CaC2 2%	2.4	1.1	3.2
CaC2 3%	1.7	1.2	1.1
CaC2 4%	1.3	1.2	0.9
CaC2 5%	3.0	1.4	0.6

3. TEST RESULTS

Table -1: Density comparison

Mix	Density In Kg/m ³		
	10M	10MS	12M
CaC2 1%	1292.39	1253.64	1268.22
CaC2 2%	1277.94	1384.83	1292.52
CaC2 3%	1394.55	1501.45	1482.02
CaC2 4%	1384.84	1326.52	1418.86
CaC2 5%	1370.26	1540.32	1496.59

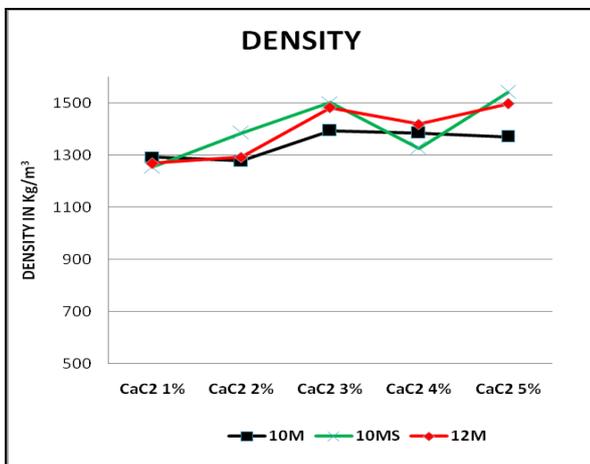


Chart -1: Density Comparison

Table -2: Water absorption comparison

Mix	Water Absorption in Percentage		
	10M	10MS	12M
CaC2 1%	8.52	3.79	7.25
CaC2 2%	8.05	3.50	7.5
CaC2 3%	11.15	3.55	9.17
CaC2 4%	12.85	4.2	10.28
CaC2 5%	13.47	4.96	12.65

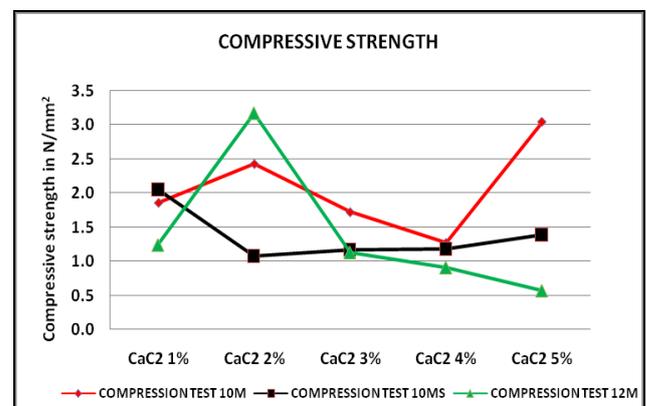


Chart -3: Compressive Strength Comparison

4. CONCLUSIONS

- The minimum density is given in CaC2 1% in all mixes except for CaC2 2% in 10M cubes. The minimum density is near the value 1300kg/m³.
- The percentage of water absorption goes on increasing with increase in percentage of CaC2. 10M cubes show more water absorption than 10MS and 12M cubes.
- Low water absorption is shown by 12M which corresponds to maximum density of 12M cubes.
- The water absorption does not exceed 14% for all cubes.

- The maximum compressive strength is given by CaC2 2% of 12M cubes. by CaC2 5% of 10M cubes.
- Except CaC2 2% of 12M, other cubes of 12M show lower value than 10M and 10MS cubes.
- Except CaC2 1% of 10M, other cubes of 10M show higher value than 10MS and 12M cubes.

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