

Analysis of Thermal Efficiency of Bio Ethyl Ester of Karanja, Jatropha and Kusum in Compression Ignition Engine

Subhash Gautam¹, Mahesh Sanadaya²

¹Student, M-Tech, Mechanical Engineering, SIT, Mathura, Uttar Pradesh, India

²Assistant Professor, Department of Mechanical Engineering, SIT, Mathura, Uttar Pradesh, India

Abstract - Biodiesel is derived from triglycerides by transesterification has attracted considerable attention during the past decade as a renewable, biodegradable and nontoxic fuel. Several processes for biodiesel fuel production have been developed. In India, non-edible oils like Jatropha oil, Karanja oil and Kusum oil are available in abundance, which can be converted to biodiesel. In the present studies, biodiesel has been prepared from Jatropha, Karanja and Kusum oil. It can be converted to biodiesel by transesterification process. The density of ethyl esters is found to be comparable with that of diesel fuel

Key Words: Diesel Engine, Bio-Diesel, Jatropha Oil, Kusum Oil, Karanja Oil.

1. INTRODUCTION

India's demand for diesel fuels is roughly six times that of gasoline hence seeking alternative to mineral diesel is a natural choice. Alternative fuels should be easily available at low cost, be environment friendly and fulfill the energy security needs without sacrificing engine operational performance. For the developing countries, fuels of bio origin provide a feasible solution to the twin crises of fossil fuel depletion and environmental degradation. Now biofuels are getting a renewed attention because of global stress on reduction of greenhouse gases and clean environment. The fuels of bio-origin may be alcohol, vegetable oils, biomass and biogas. Some of the fuels can be used directly while others need to be formulated to bring the relevant properties close to conventional fuels. Vegetable oils have comparable energy density, cetane number, heat of vaporization, and stoichiometric air/fuel ratio with mineral diesel fuel [1]. Biodiesel has been touted far and wide for its renewable properties. Instead of making a fuel from a finite resource such as crude oil, Biodiesel can be produced from renewable resources such as organic oils, fats. This means that it can be made from things that can be regrown, reproduced, and reused. So, if you need more, you can just grow another crop of seeds for the oil.

When Biodiesel is made from organic oils such as jatropha, Karanja, kusum, or other domestically grown seed crops, it helps the farming community out. Because the oil used to make Biodiesel is "domestically grown", it keeps the money flowing to those that "grow" the feedstock. This continues to

help out the renewable aspect of Biodiesel because this means more seed crops can be grown by local farmers. When Biodiesel is used in place of petro diesel, it reduces the amount of crude oil used up. This means that it helps to reduce our dependence on a limited resource and increases our use of renewable resources. We think that's a great step toward reducing our dependence on a fuel that may not be around forever.

1.1 Karanja Oil

Pongamia pinnata is commonly known as karanja in the north and eastern states of India. A single tree yields 9–90 kg seed per tree, indicating a yield potential of 900–9 000 kg seed/ha (assuming 100 trees/ha), 25% of which might be rendered as oil. In general, Indian mills extract 24%–27.5% oil, and the village crushers extract 18%–22% oil [1].



Fig -1: Leaves and Fruits of Karanja

1.2 Jatropha Oil

Jatropha curcas also known as ratanjot. The leaves are green, 10-15 cm in width and length. The shape of fruits having an ellipsoidal. Each fruit is about 40 mm long and contains three seeds. It takes three to four months after the flowering for the seeds to mature. The seeds are black, measuring on average 18 mm in length, 12 mm in width, and 10 mm in thickness. The seeds weigh between 0.5 and 0.8 grams and the average number of seeds per kilo is 1375 seeds. Oil content of the seeds range from 32 to 40 percent [3].



Fig – 2: Fruits and seeds of Jatropha

1.3 Kusum Oil

The botanical name of kusum is *Schleichera olcosa* and the potential of kusum oil is 66000 tons per year in India, out of which 4 000 to 5 000 tones are collected. It is a medium to large sized, dense tree growing to 35 to 45 feet in height. It mainly occurs in sub-Himalayan tracts in the north, central parts of eastern India. The flowers come from February to April and yields fruit in June and July. The fruits are smooth, hard skin berries contains one or two irregularly ellipsoidal slightly compressed seeds. The brown seed coat is brittle and breaks at a slight pressure to expose a 'U-shaped kernel. The oil content is 51%–62% but the yields are 25%–27% in village gains (oil mills) and about 36% oil in expellers [1]. The weight of 1000 seeds is 500-700 g. *Schlichera oleosa* evergreen trees with the height up to 30 m and the girth up to 3 m. The leaves are paripinnate, 20-40 cm long [7].



Fig – 3: Tree and Fruits of Kusum

2. MATERIALS AND METHODS

Bio-diesel is fatty acid ethyl ester made from virgin or used vegetable oils (both edible & non-edible) and animal fats. The main commodity sources for bio-diesel in India can be

non-edible oils obtained from plant species such as *Schleichera olcosa* (kusum), *Pongamia Pinnata* (Karanja), *Jatropha curcas* (Ratanjyot) etc.

2.1 Preparation of biodiesel from Karanja Oil

Biodiesel is a methyl ester formed by a process called Transesterification. Oil can be extracted from seeds of Karanja with a simple mill (as used for mustard seeds). The Karanja oil reacted with Methanol in the presence of a catalyst to yield methyl esters and glycerol. Sodium Hydroxide (NaOH) and Potassium Hydroxide (KOH) are commonly used catalyst. Heat the oil at 60°C. Dissolve Sodium Hydroxide tablets in methanol to make solution. The quantity of NaOH and methanol should be 2.5% and 10%, respectively of the total quantity of Karajna oil. After mixing this solution into hot Karanja oil, the solution should be stir for 5 to 7 minutes. Then keep the solution undisturbed at least 4 hours. Glycerol being heavy will slowly settle down at the bottom and biodiesel can be easily separated from the top. To strain the impurities like sodium, give this oil 2-3 washing with water. Add water to the oil and after 5 minutes collect the oil floating on surface. Repeat this process and then finally heat the oil to evaporate the water. The biodiesel is no now ready to use [2].

2.2 Preparation of biodiesel from Jatropha Oil

Biodiesel was prepared with methanol and ethanol each with different reaction conditions. With methanol, the experiment was conducted with optimum molar ratio (6:1) keeping the catalyst concentration (1% NaOH), reaction temperature (65°C) and reaction time (1 hour). With ethanol, the experiment was conducted with optimum molar ratio (8:1) keeping the catalyst concentration (1% KOH), reaction temperature (70°C) and reaction time (3½ hour) [4,6]. The required amount of jatropha oil was filtered, measured with measuring cylinder and then it was poured into the three necked round-bottomed flask. The jatropha oil was heated to the required temperature by using the electric mental. Alkoxide solution was prepared while the jatropha oil was heated. The prepared alkoxide solution was introduced into the reaction vessel and it was mixed vigorously during the reaction. When the required reaction period reached, the reaction was stopped, and the mixture was settled in the separating funnel for 12 hours or overnight. After the mixture was settled for 12 hours, the mixture was separated into two layers. The bottom layer is crude glycerine and it can be drawn off simply from the bottom of the separating funnel. The biodiesel layer was purified by washing with warm water to remove methanol, residual catalyst and soaps. Before washing process, the pH of the biodiesel layer was measured and phosphoric acid was added to the biodiesel layer to neutralize the catalyst residue. After neutralization process, the washing process of biodiesel was started. During the washing process, gentle agitation is required to avoid the emulsion. After separation

of the layer for 30 minutes, the wash water layer was drained off from the bottom of the separating funnel. The washing process was repeated until the ester layer became clear. After the washing process, it was required to measure the pH of the biodiesel layer. When the pH of the biodiesel layer reached 7, the washing process was completed. After washing process, the biodiesel was introduced to the sand filter and salt filter. The end product, biodiesel was obtained as a clear amber-yellow liquid with a viscosity similar to that of petrodiesel [5,6].

2.3 Preparation of biodiesel from Kusum Oil

Bio-diesel is produced by transesterification which is a process of using either ethanol or methanol, in the presence of a catalyst, such as potassium hydroxide or NaOH, to chemically break the molecule of an oil or fat into an ester and glycerol. This process is a reaction of the oil with an alcohol to remove the glycerine, which is a byproduct of biodiesel production. The reactor for producing biodiesel from Schlichera oleosa oil is a small batch type reactor. The first we are taken pretreatment oil temperature is 50-55°C. The products of the first stage pretreatment oil are used as the input of the alkaline Transesterification process. A molar ratio of 6:1 and 10grms by weight of potassium hydroxide (KOH) is found to give the maximum ester yield. The reaction time is maintaining 2hr at 60°C. After the reaction is completed, the products are allowed to separate into two layers. The lower layer contains impurities and glycerol. This top layer (ester) is separated and purified using distilled water. Hot distilled water (20% by volume) is sprayed over the ester and stirred gently and allowed to settle in the separating funnel. The lower layer is discarded and upper layer (purified biodiesel) is separated. Transesterification of the oil produces methyl esters (biodiesel) and glycerol. The methyl ester layer is a light yellow liquid that is on top or bottom of the glycerol layer, which is dark brown in color. The mixtures may be kept overnight and allowed to separate by gravity. Otherwise, the methyl ester is separated from the glycerol and washed with water and acetic acid until the washing water is neutral. The methyl ester is then dried by heating. Materials and energy flows for the transesterification of Schlichera oliosa seed oil into biodiesel [7].

3. EXPERIMENT

Engine Specification

Engine	Kirloskar (AV1-5.0)
Make & Type	Single cylinder, Four Stroke, Water Cooled, Vertical Engine
Bore	80 mm
Stroke	110 mm
Displacement volume	553 cc

Compression ratio 16.5:1
Rated output at 1500 RPM 3.7 KW (5.0 H.P.)

3.1 Formula Used

$$\text{Break Power} = \frac{2\pi NT}{60 \times 1000} \text{KW}$$

Where

$$T = g(S_1 - S_2)L \text{ N-m}$$

N = Engine speed in RPM

T = Torque in N-m

S1&S2 = Spring balance

$$\text{Specific Fuel Consumption} = \frac{m}{B.P} \text{ gm/KW-h}$$

Where

m = Fuel in gm

B.P = Break power

$$\text{Fuel Consumption} = \frac{m}{t} \text{ gm/sec}$$

Where

t = time in sec

m = Fuel in gm

$$\text{Indicated Power} = \frac{m \times cv}{t \times 1000} \text{ KW}$$

Where

cv = Calorific value of fuel used in KJ/Kg

m = Fuel in gm

t = time in sec

$$\text{Thermal Efficiency} = \frac{B.P \times t}{m \times cv} \times 100 \%$$

Where

B.P = Break power

t = time in sec

m = Fuel in gm

cv = Calorific value of fuel used in KJ/Kg



Fig - 4: Laboratory experimental setup

Table -1: Fuel Specification

Fuel	Density (gm/cm ³)	Calorific value (KJ/Kg)
Diesel [8]	0.8227	41840
Jatropha [9]	0.8667	39230
Karanja [10,11]	0.8700	36577
Kusum [1,12]	0.8620	35000

4. RESULT AND CONCLUSIONS

The Performance of Biodiesel blends at different loads is different but some blends give better performance than Diesel at same load.

Karanja biodiesel is efficient than diesel at no load condition at D90:B10 ratio.

Jatropha biodiesel is efficient than diesel on 0.7 KW, 2.1 KW at D80:B20, D90:B10, ratio respectively. And at 2.8 KW it is better than karanja and kusum biodiesel but lesser than diesel.

Kusum biodiesel is efficient than diesel on 1.4 KW, 3.5 KW at D60:B40, D90:B10 ratio respectively

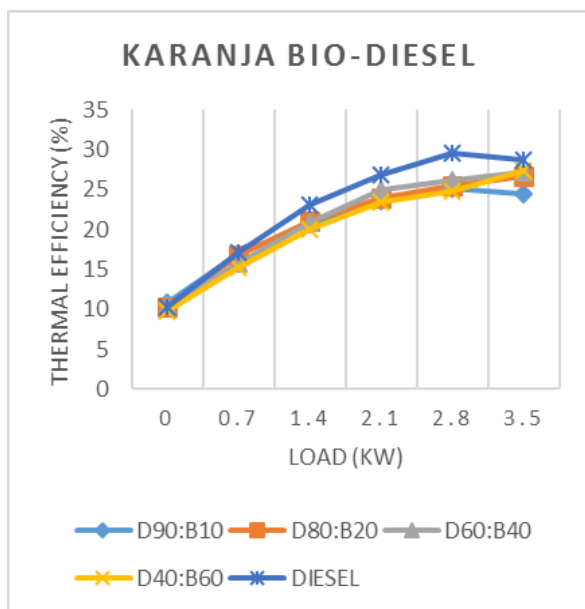


Chart - 1: Performance of thermal efficiency with load (KW) of Karanja biodiesel blends.

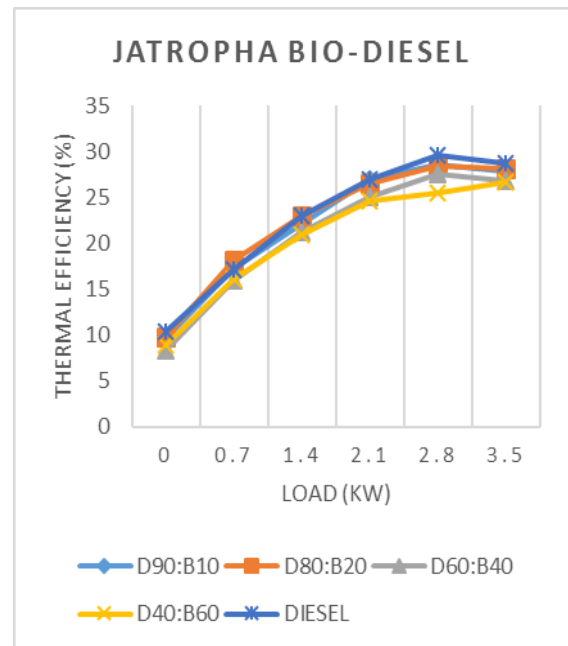


Chart - 2: Performance of thermal efficiency with load (KW) of Jatropha biodiesel blends.

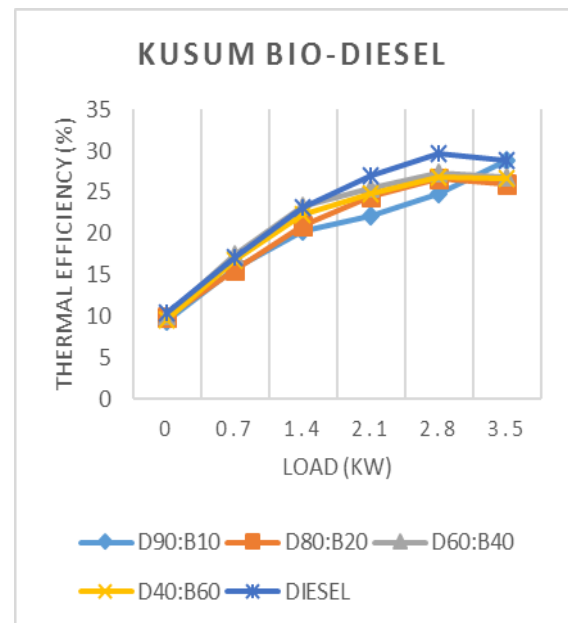


Chart - 3: Performance of thermal efficiency with load (KW) of Kusum biodiesel blends.

Thermal efficiencies of jatropha, karanja and kusum biodiesel shows their performance according to blends ratio with diesel at different loads (in KW) in charts.

Table -2: Density and calorific value of Karanja biodiesel

Fuel	Karanja	
Fuel Bled	Density (gm/cm ³)	Calorific value (KJ/Kg)
D90:B10	0.82743	41313.7
D80:B20	0.83216	40787.4
D60:B40	0.84162	39734.8
D40:B60	0.85108	38682.2

Above table shows the density and calorific value of Karanja biodiesel at different blend ratio.

Table -3: Density and calorific value of Jatropa biodiesel

Fuel	Jatropa	
Fuel Bled	Density (gm/cm ³)	Calorific value (KJ/Kg)
D90:B10	0.8291	41579
D80:B20	0.8315	41318
D60:B40	0.8403	40796
D40:B60	0.8491	40274

Above table shows the density and calorific value of jatropa biodiesel at different blend ratio.

Table -4: Density and calorific value of Kusum biodiesel

Fuel	Kusum	
Fuel Bled	Density (gm/cm ³)	Calorific value (KJ/Kg)
D90:B10	0.82663	41156
D80:B20	0.83056	40472
D60:B40	0.83842	39104
D40:B60	0.84628	37736

Above table shows the density and calorific value of kusum biodiesel at different blend ratio.

Table -5: Comparison of thermal efficiency between biodiesel blends and diesel

Load (KW)	Biodiesel	Best Ratio	Best Thermal Efficiency	
			Diesel	Biodiesel
0	Karanja	D90:B10	10.3549	10.7916
0.7	Jatropa	D80:B20	17.1321	18.1578
1.4	Kusum	D60:B40	23.0588	23.2927
2.1	Jatropa	D90:B10	26.9260	26.9784
2.8	Jatropa	D90:B10	29.6103	28.5884
3.5	Kusum	D90:B10	28.7739	28.8247

Above table shows the comparison of thermal efficiency between biodiesel blends and diesel with load and ratio.

CONCLUSION

The main aim of the present experiment was to use the non-edible oil like jatropa, karanja and kusum oil in diesel engines. We reduced the oil density close to that of diesel without using any external power source and evaluating the performance of engine with the modified oils. The density of jatropa, karanja and kusum oil was reduced by transesterification process. It was found that in the above cases the density was close to that of diesel –which would be suitable for the engines. The performance tests were conducted with diesel, From the experimental investigation it was concluded that the performance of jatropa, karanja and kusum oil is similar to that of diesel, without any operational difficulties. In comparison to each oil defined below.

Jatropa biodiesel shows best performance between 0 KW – 1.4 KW, 3.5 KW with D80:B20 ratio and on 2.1 KW-2.8 KW with D90:B10 ratio.

Kusum biodiesel shows best performance between 0 KW-2.8 KW with D60:B40 ratio and on 3.5KW with D90:B10 ratio.

Karanja biodiesel is one which is give its performance increasing in order with increasing loads and blends ratio. It gives best at 0 KW-0.7 KW with D90:B10 ratio, at 1.4 KW with D80:B20 ratio, at 2.1 KW-2.8 KW with D60:B40 ratio and at 3.5 KW with D40:B60 ratio.

REFERENCES

- [1] S K Acharya, A K Mishra, M Rath, C Nayak, "Performance Analysis of Karanja and Kusum Oils as Alternative Bio-Diesel Fuel in Diesel Engine," Int J Agric & Biol Eng June, 2011 Vol. 4 No.2 issn.1934-6344.
- [2] R.S. Kureel, C.B.Singh, A.K.Gupta,Ashutosh Pandey, "Karanja a potential source of biodiesel, national oilseeds & vegetable oils," Development board ministry of agriculture, govt. Of india, may 2008.
- [3] Lisa Axelsson Maria Franzén, "Performance of Jatropa Biodiesel Production and Its Environmental and Socio-Economic Impacts," Department of Energy and Environment Chalmers University of Technology Göteborg, Sweden, 2010 Report No. FRT 2010:06.
- [4] P.Suresh Kumar, Ramesh Kumar Donga, P.K. Sahoo, "Experimental comparative study between performance and emissions of jatropa biodiesel and diesel under varying injection pressures," International Journal of Engineering Sciences & Emerging Technologies, August 2012. Volume 3, ISSN: 2231 – 6604.
- [5] V. B. Shambhu1, T. K. Bhattacharya, L. K. Nayak and S. Das Senior Scientist, "Studies on Characterization of Raw Jatropa Oil and its Biodiesels with Relevance of Diesel,"

International Journal of Emerging Technology and Advanced Engineering April Volume 3, 2013 ISSN 2250-2459.

- [6] Tint Tint Kywe, Mya Mya Oo, "Production of Biodiesel from Jatropha Oil (Jatropha curcas) in Pilot Plant." World Academy of Science, Engineering and Technology 26 2009.
- [7] Mallela Gandhi, N. Ramu and S. Bakkiya Raj, "Methyl ester production from schlichera oleosa," IJPSR, 2011; Vol. 2(5): 1244-1250 ISSN: 0975-8232.
- [8] High Speed Diesel / Gas Oil Indian Oil Diesel meets the requirements of IS 1460:2005 (5th revision).
- [9] Wilson Parawira, "Biodiesel production from Jatropha curcas: A Review," Scientific Research and Essays Vol. 5(14), pp. 1796-1808, 18 July, 2010 ISSN 1992-2248.
- [10] Karikalan and M. Chandrasekaran, "Karanja oil biodiesel: a potential substitution for diesel fuel in diesel engine without alteration." ARPN Journal of Engineering and Applied Sciences VOL. 10, NO. 1, January 2015 ISSN 1819-6608.
- [11] Bobade S.N. and Khyade V.B. Indian "Biodiesel Corporation, Baramati Detail study on the Properties of Pongamia Pinnata (Karanja) for the Production of Biofuel," INDIA Research Journal of Chemical Sciences Vol. 2(7), 16-20, July (2012) ISSN 2231-606X.
- [12] S.K.Acharya, M.K.Mohanty R.K.Swain, "Kusum Oil as a Fuel for Small Horse Power Diesel Engine." International Journal of Engineering and Technology Vol. 1, No.3, August, 2009 ISSN: 1793-8236.