

# To Obtain Pyrolysis Oil from Waste Plastics of Different Grades & To Determine Its Calorific Value

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**Abstract** - The global demand for energy has been ever increasing in recent years. The demand, especially from liquid fuels is very high and the limited resources of fuel production have created bottleneck leading to an energy crisis. This has led to exploring other resources for fuel production, one of which is plastic. Being a non-degradable source, plastics disposed of in the open environment as wastes pose a threat to the environment. Most of the waste plastics end up in landfills. It can instead be used as a source for making fuel.

The work describes a process to use the waste plastic to synthesize potential fuel called 'Pyrolysis Oil' by Pyrolysis process. The obtained oil from different grades of waste plastics is analyzed so as to confirm its use as fuel.

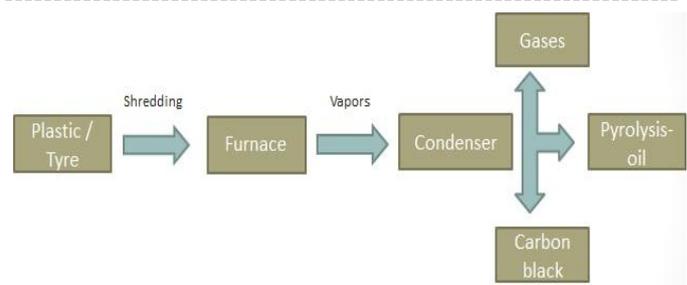
The paper describes the process of extraction of pyrolysis oil from waste polymers of different grades by fabricating a furnace to carry out pyrolysis.

**Key Words:** Waste Plastic, Pyrolysis, Pyrolysis Oil, Calorific Value.

## 1. INTRODUCTION

Plastic waste management is a big issue in India. As per the Central Pollution Control Board (CPCB), 5.6 million tons of plastic and rubber waste is generated annually and approximately only 60% of collected plastic and rubber waste is re- cycled in India. [1]Tons of Plastic waste is dumped on land and huge amounts are disposed of into the water bodies. This plastic and rubber waste could be used to produce fuel. Pyrolysis of waste plastic and rubber could provide a better way to dispose of the waste materials.

A thermochemical decomposition of organic material at elevated temperatures in the absence of oxygen is called Pyrolysis. Pyrolysis of organic substances produces gas and liquid products 'biofuels' and leave a solid residue richer in carbon content.



The project is thus selected with an objective of using this non degradable waste plastic and rubber as a source to extract fuel which after analysis can be used as an alternative source of energy.

## 2. LITERATURE REVIEW

Manish Chand Sharma [2] issued a paper in 2013 on "Production of alternative fuel from waste oil and comparison with fresh diesel". This research paper compares the blend of fresh diesel and diesel obtained from pyrolysis of used engine oils with conventional diesel oil.

John William Bordynuik [3], issued a paper in March 2013 "Performance and Emission Evaluation of Blends of Diesel fuel with Waste Plastic Oil in a Diesel Engine", it studied Viable Production of Diesel from non- recyclable waste plastics.

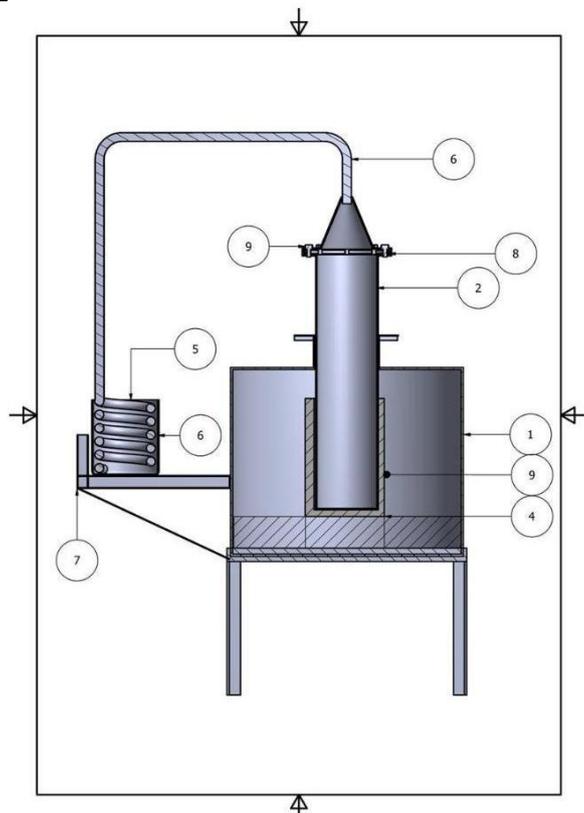
C. Wongkhorsub et.al. [4] published a paper in July 2013 titled "A Comparison of the Use of Pyrolysis Oils in Diesel Engine". This research describes a comparison of the use of pyrolysis oils which are the tire pyrolysis oil, plastic pyrolysis oil and diesel oil in the assessment of engine performance, and feasibility analysis. Pyrolysis oils and waste plastic are studied to apply with one cylinder multipurpose agriculture diesel engine. Thus, a comparison between the two is studied.

## 3. MODELLING AND FABRICATION OF SETUP

The experimental setup was modelled in Solid Works software with certain specifications. The need of different components was determined by the layout and was modelled based on the holding capacity of the heating chamber. The parts have been marked according to the parts list.

**Table -1:** Parts list for the setup

Sr. No	Part Name	Quantity
1	Frame	1
2	Heating Chamber	1
3	Furnace (Heating coil of 3kW)	1
4	Insulations	1
5	Condenser	1
6	Tank	1
7	Bracket	1
8	Gasket	1
9	Nut, Bolt & Washer	8



**Fig -1:** Proposed assembly of experimental setup

The basic components of the setup include a heating unit, a cooling unit and collection unit. The model of the proposed assembly helps to give a better idea and understanding of the setup and helps in the fabrication of components.

### 3.1 Furnace

The material used for the fabrication of major components is mild steel. Mild steel has a carbon content ranging from 0.15% to 0.30%. It has properties suitable for fabrication and is available easily.



**Fig -2:** Furnace

A 3 kW electric coil was selected for the purpose of attaining the required temperature in the heating chamber. The coil has been wrapped around a ceramic block of dimensions 170 mm x 170 mm x 300 mm. The coil was sealed in its place with the help of furnace cement. The figure shows the electric coil enclosed around the block and secured in its place by furnace cement. Glass wool was used for insulation around the furnace.

### 3.2 Heating Chamber

The heating chamber is the most important component of the setup as it has to sustain the heat generated by the furnace and also be free of any leakage. The heating chamber basically consists of two main parts, the conical section with a flange and the pipe with a flange. The pipe has an outer diameter of 14 cm and a length of 50 cm. The thickness of the material used is 4.5 mm.



**Fig -3:** Heating Chamber

A high temperature gasket is selected to be placed in between these flanges to ensure an air-tight seal. The gasket is circular in shape with an inside diameter of 15 cm and outside diameter of 16 cm.

### 3.3 Heat Exchanger

A helical coil heat exchanger consists of a pipe passing through a tank containing water. A standard mild steel pipe of 1 inch diameter and 2.8 m length was bent to fabricate a helical coil pipe using hot working.



Fig -4: Heat Exchanger

A stainless steel utensil of 28 cm diameter is used for the purpose of holding water across the helical coil. The tank is welded to the coil outlet and a water tight seal is applied across the weld.

### 3.4 Frame



Fig -5: Frame

The circular frame of 52 cm diameter and 42 cm length is prepared by rolling a sheet of metal and welding its ends. The bottom is welded with circular shape metal sheet of 3 mm thickness. The frame houses the furnace and the chamber and also provides thermal insulation.

## 4. ASSEMBLY OF PYROLYSIS SETUP

Assembly of the setup follows the fabrication and testing of the components required for the process. The stand is placed where the setup is to be installed so as to conduct the experiment. The experiment is performed outdoors for safety. The frame is carefully placed on the stand. The next step involves carefully installing the furnace and the insulation bricks. Then carefully place the furnace in the center. The wires of the electric furnace are sealed in insulating material. Bricks are placed in the rest of the space surrounding the furnace. This ensures minimum heat loss from the furnace in all directions except the top. The frame

lid is then used to cover the furnace inside the frame. The lid prevents heat loss. The penultimate step is inserting the heating chamber through the lid into the furnace. The final step of the assembly is placing the helical coil heat exchanger assembly on the stand.



Fig -6: Final setup

For performance of the experiment, the material selected is first added to the heating chamber. The next step is placing the heat sealing gasket between the flanges and fastening the flanges using nut bolting. The fasteners are tightened so that the gasket properly seals the flanges. No air leakage is tolerated as this may result into combustion of the polymers being used. The empty water tank is filled with water before starting the test. The water is filled so that the major part of the coil is immersed under water. This helps to increase the effectiveness in heat exchange. After final inspection the electric supply to the furnace is switched on and the experiment begins.

A stopwatch is used so as to keep a track of observations with respect to time. After switching on the apparatus, approximately 15 minutes later it is observed that furnace starts heating considerably. This can be said as the hot fumes are seen rising up from the little gaps in the frame and the heating chamber.

Approximately 40 minutes after the start it is observed that milky fumes are obtained from the outlet. Also the heat sealing gasket is seen to emit some fumes as it heats up. However these observations last for a few minutes until the entire setup attains high temperature (400°C+).

Using 2.6 kg of plastic in the experimentation 85 minutes after the start of the experiment it is observed that drops of oil start falling from the outlet. A funnel and a flask for oil collection are placed below the outlet.

## 5. RESULTS AND DISCUSSION

The setup is fabricated for performing experiments with different grades of plastic. Experiment is conducted with waste plastic from injection moulding (Grade 5), Waste Plastic - Grade 2 & PVC Pipes and following results were obtained.

**Table -2:** List of experimental results

Plastic	Input (Kg)	Output (L)
Waste plastic from injection molding (Grade 5)	2.6	3.1
Waste plastic (Grade 2 )plastic	2.7	3.0
PVC Pipes	1.0	0.2



**Fig-7:** Injection moulding Grade 5



**Fig-8:** Oil obtained from injection moulding Grade 5



**Fig-9:** Waste Plastic - Grade 2



**Fig-10:** Oil obtained from Waste Plastic - Grade 2



**Fig-11:** PVC Pipes



**Fig-12:** Oil obtained from PVC Pipes

The former 2 samples of oil have been analysed and tested for their thermal and fuel properties. The testing of the fuel samples have been carried out by HPCL R&D Laboratories, Turbhe, Navi Mumbai.

**Table 2:** Calorific value of the fuels

Fuels	GCV (kCal/Kg)	NCV (kCal/Kg)
Sample 1 (Grade 5 – PP)	10885	10202
Sample 2 (Grade 2 – HDPE)	10972	10276
Diesel Oil	10700	10366

## 6. CONCLUSIONS

The Pyrolysis process was carried out with Grade 5, Grade 2 and PVC type of plastic materials. Quantity and quality of oil can be varied by using different quantity of plastics of various grades. The properties are found close to Diesel and hence

pyrolysis oil can be used with blends of Diesel oil after further treatment. The obtained fuel is a raw mixture of hydrocarbons. It can be fractionally distilled to obtain the desired composition of oil.

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